

Fold  
↓**I POINT QUESTION**

Convert 12 grams of C to moles of C

$$12 \text{ grams C} \left( \frac{1 \text{ mole C}}{12.01 \text{ grams C}} \right)$$

**ANSWER: 1.0 mole of C****I POINT QUESTION**

Convert 34 grams of O to moles of O

$$34 \text{ grams O} \left( \frac{1 \text{ mole O}}{16 \text{ grams O}} \right)$$

**ANSWER: 2.1 moles of O****I POINT QUESTION**

Convert 18 grams of N to moles of N

$$18 \text{ grams N} \left( \frac{1 \text{ mole N}}{14.01 \text{ grams N}} \right)$$

**ANSWER: 1.3 moles of N****I POINT QUESTION**

Convert 78 grams of Ne to moles of Ne

$$78 \text{ grams Ne} \left( \frac{1 \text{ mole Ne}}{20.18 \text{ grams Ne}} \right)$$

**ANSWER: 3.9 moles of Ne**

Fold at the --- line  
Cut at the —— lines

**1 POINT QUESTION**

Convert 15 grams of H to moles of H

$$15 \text{ grams H} \left( \frac{1 \text{ mole H}}{1.008 \text{ grams H}} \right)$$

**ANSWER:** 15 moles of H

**1 POINT QUESTION**

Convert 62 grams of He to moles of He

$$62 \text{ grams He} \left( \frac{1 \text{ mole He}}{4.003 \text{ grams He}} \right)$$

**ANSWER:** 15 moles of He

**1 POINT QUESTION**

Convert 89 grams of Xe to moles of Xe

$$89 \text{ grams Xe} \left( \frac{1 \text{ mole Xe}}{131.29 \text{ grams Xe}} \right)$$

**ANSWER:** 0.68 moles of Xe

**1 POINT QUESTION**

Convert 65 grams of Pd to moles of Pd

$$65 \text{ grams Pd} \left( \frac{1 \text{ mole Pd}}{106.42 \text{ grams Pd}} \right)$$

**ANSWER:** 0.61 moles of Pd

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 25 grams of V to moles of  
V

$$25 \text{ grams } V \left( \frac{1 \text{ mole } V}{50.94 \text{ grams } V} \right)$$

***ANSWER: 0.49 moles of V***

***1 POINT QUESTION***

Convert 72 grams of W to moles of  
W

$$72 \text{ grams } W \left( \frac{1 \text{ mole } W}{183.85 \text{ grams } W} \right)$$

***ANSWER: 0.39 moles of W***

***1 POINT QUESTION***

Convert 63 grams of Mg to moles of  
Mg

$$63 \text{ grams } Mg \left( \frac{1 \text{ mole } Mg}{24.31 \text{ grams } Mg} \right)$$

***ANSWER: 2.6 moles of Mg***

***1 POINT QUESTION***

Convert 32 grams of Be to moles of  
Be

$$32 \text{ grams } Be \left( \frac{1 \text{ mole } Be}{9.012 \text{ grams } Be} \right)$$

***ANSWER: 3.6 moles of Be***

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 64 grams of Cs to moles of Cs

$$64 \text{ grams Cs} \left( \frac{1 \text{ mole Cs}}{132.9 \text{ grams Cs}} \right)$$

***ANSWER: 0.48 moles of Cs***

***1 POINT QUESTION***

Convert 86 grams of K to moles of K

$$86 \text{ grams K} \left( \frac{1 \text{ mole K}}{39.1 \text{ grams K}} \right)$$

***ANSWER: 2.2 moles of K***

***1 POINT QUESTION***

Convert 32 grams of Cr to moles of Cr

$$32 \text{ grams Cr} \left( \frac{1 \text{ mole Cr}}{52.0 \text{ grams Cr}} \right)$$

***ANSWER: 0.62 moles of Cr***

***1 POINT QUESTION***

Convert 98 grams of Au to moles of Au

$$98 \text{ grams Au} \left( \frac{1 \text{ mole Au}}{197 \text{ grams Au}} \right)$$

***ANSWER: 0.50 moles of Au***

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 12 grams of Al to moles of Al

$$12 \text{ grams Al} \left( \frac{1 \text{ mole Al}}{26.98 \text{ grams Al}} \right)$$

***ANSWER: 0.44 moles of Al***

***1 POINT QUESTION***

Convert 52 grams of B to moles of B

$$52 \text{ grams B} \left( \frac{1 \text{ mole B}}{10.811 \text{ grams B}} \right)$$

***ANSWER: 4.8 moles of B***

***1 POINT QUESTION***

Convert 71 grams of Pb to moles of Pb

$$71 \text{ grams Pb} \left( \frac{1 \text{ mole Pb}}{207.2 \text{ grams Pb}} \right)$$

***ANSWER: 0.34 moles of Pb***

***1 POINT QUESTION***

Convert 54 grams of Si to moles of Si

$$54 \text{ grams Si} \left( \frac{1 \text{ mole Si}}{28.09 \text{ grams Si}} \right)$$

***ANSWER: 1.9 moles of Si***

Fold at the --- line  
Cut at the —— lines

**1 POINT QUESTION**

Convert 52 grams of  $\text{FeCl}_3$  to moles of  $\text{FeCl}_3$

$$52 \text{ grams } \text{FeCl}_3 \left( \frac{1 \text{ mole } \text{FeCl}_3}{162.2 \text{ grams } \text{FeCl}_3} \right)$$

**ANSWER: 0.32 moles of  $\text{FeCl}_3$**

**1 POINT QUESTION**

Convert 43 grams of  $\text{Fe}_2\text{O}_3$  to moles of  $\text{Fe}_2\text{O}_3$

$$43 \text{ grams } \text{Fe}_2\text{O}_3 \left( \frac{1 \text{ mole } \text{Fe}_2\text{O}_3}{159.69 \text{ grams } \text{Fe}_2\text{O}_3} \right)$$

**ANSWER: 0.27 moles of  $\text{Fe}_2\text{O}_3$**

**1 POINT QUESTION**

Convert 14 grams of  $\text{MgO}$  to moles of  $\text{MgO}$

$$14 \text{ grams } \text{MgO} \left( \frac{1 \text{ mole } \text{MgO}}{40.305 \text{ grams } \text{MgO}} \right)$$

**ANSWER: 0.35 moles of  $\text{MgO}$**

**1 POINT QUESTION**

Convert 72 grams of  $\text{MgCl}_2$  to moles of  $\text{MgCl}_2$

$$72 \text{ grams } \text{MgCl}_2 \left( \frac{1 \text{ mole } \text{MgCl}_2}{95.211 \text{ grams } \text{MgCl}_2} \right)$$

**ANSWER: 0.76 moles of  $\text{MgCl}_2$**

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 67 grams of H<sub>2</sub> to moles of H<sub>2</sub>

$$67 \text{ grams H}_2 \left( \frac{1 \text{ mole H}_2}{2.016 \text{ grams H}_2} \right)$$

***ANSWER: 33 moles of H<sub>2</sub>***

***1 POINT QUESTION***

Convert 23 grams of Li<sub>3</sub>PO<sub>4</sub> to moles of Li<sub>3</sub>PO<sub>4</sub>

$$23 \text{ grams Li}_3\text{PO}_4 \left( \frac{1 \text{ mole Li}_3\text{PO}_4}{115.797 \text{ grams Li}_3\text{PO}_4} \right)$$

***ANSWER: 0.20 moles of Li<sub>3</sub>PO<sub>4</sub>***

***1 POINT QUESTION***

Convert 51 grams of H<sub>3</sub>PO<sub>4</sub> to moles of H<sub>3</sub>PO<sub>4</sub>

$$51 \text{ grams H}_3\text{PO}_4 \left( \frac{1 \text{ mole H}_3\text{PO}_4}{97.998 \text{ grams H}_3\text{PO}_4} \right)$$

***ANSWER: 0.52 moles of H<sub>3</sub>PO<sub>4</sub>***

***1 POINT QUESTION***

Convert 42 grams of ZnS to moles of ZnS

$$42 \text{ grams ZnS} \left( \frac{1 \text{ mole ZnS}}{97.454 \text{ grams ZnS}} \right)$$

***ANSWER: 0.43 moles of ZnS***

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 31 grams of O<sub>2</sub> to moles of O<sub>2</sub>

$$31 \text{ grams } O_2 \left( \frac{1 \text{ mole } O_2}{32 \text{ grams } O_2} \right)$$

***ANSWER: 0.97 moles of O<sub>2</sub>***

***1 POINT QUESTION***

Convert 57 grams of ZnO to moles of ZnO

$$57 \text{ grams } ZnO \left( \frac{1 \text{ mole } ZnO}{81.39 \text{ grams } ZnO} \right)$$

***ANSWER: 0.70 moles of ZnO***

***1 POINT QUESTION***

Convert 0.35 moles of SO<sub>2</sub> to grams of SO<sub>2</sub>

$$0.35 \text{ moles } SO_2 \left( \frac{64.064 \text{ grams } SO_2}{1 \text{ mole } SO_2} \right)$$

***ANSWER: 22 grams of SO<sub>2</sub>***

***1 POINT QUESTION***

Convert 0.23 moles of HCl to grams of HCl

$$0.23 \text{ moles } HCl \left( \frac{36.46 \text{ grams } HCl}{1 \text{ mole } HCl} \right)$$

***ANSWER: 8.4 grams of HCl***

Fold at the --- line  
Cut at the —— lines

**1 POINT QUESTION**

Convert 0.41 moles of MnCl<sub>2</sub> to grams of MnCl<sub>2</sub>

$$0.41 \text{ moles } \text{MnCl}_2 \left( \frac{125.84 \text{ grams } \text{MnCl}_2}{1 \text{ mole } \text{MnCl}_2} \right)$$

**ANSWER: 52 grams of MnCl<sub>2</sub>**

**1 POINT QUESTION**

Convert 0.17 moles of Cl<sub>2</sub> to grams of Cl<sub>2</sub>

$$0.17 \text{ moles } \text{Cl}_2 \left( \frac{70.9 \text{ grams } \text{Cl}_2}{1 \text{ mole } \text{Cl}_2} \right)$$

**ANSWER: 12 grams of Cl<sub>2</sub>**

**1 POINT QUESTION**

Convert 0.51 moles of H<sub>2</sub>O to grams of H<sub>2</sub>O

$$0.51 \text{ moles } \text{H}_2\text{O} \left( \frac{18.016 \text{ grams } \text{H}_2\text{O}}{1 \text{ mole } \text{H}_2\text{O}} \right)$$

**ANSWER: 9.2 grams of H<sub>2</sub>O**

**1 POINT QUESTION**

Convert 0.820 moles of Na<sub>2</sub>SiO<sub>3</sub> to grams of Na<sub>2</sub>SiO<sub>3</sub>

$$0.820 \text{ mol } \text{Na}_2\text{SiO}_3 \left( \frac{122.066 \text{ g } \text{Na}_2\text{SiO}_3}{1 \text{ mol } \text{Na}_2\text{SiO}_3} \right)$$

**ANSWER: 100. grams of Na<sub>2</sub>SiO<sub>3</sub>**

Fold at the --- line  
Cut at the —— lines

***1 POINT QUESTION***

Convert 1.2 moles of HF to grams of HF

$$1.2 \text{ moles HF} \left( \frac{20.01 \text{ grams HF}}{1 \text{ mole HF}} \right)$$

***ANSWER: 24 grams of HF***

***1 POINT QUESTION***

Convert 1.70 moles of H<sub>2</sub>SiF<sub>6</sub> to grams of H<sub>2</sub>SiF<sub>6</sub>

$$1.70 \text{ mol H}_2\text{SiF}_6 \left( \frac{144.09 \text{ g H}_2\text{SiF}_6}{1 \text{ mol H}_2\text{SiF}_6} \right)$$

***ANSWER: 245 grams of H<sub>2</sub>SiF<sub>6</sub>***

***1 POINT QUESTION***

Convert 2.3 moles of NaF to grams of NaF

$$2.3 \text{ moles NaF} \left( \frac{41.988 \text{ grams NaF}}{1 \text{ mole NaF}} \right)$$

***ANSWER: 97 grams of NaF***

***1 POINT QUESTION***

Convert 3.1 moles of NaCl to grams of NaCl

$$3.1 \text{ moles NaCl} \left( \frac{58.443 \text{ grams NaCl}}{1 \text{ mole NaCl}} \right)$$

***ANSWER: 180 grams of NaCl***